WithOnePlanet

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Lesson 1 Student worksheets Carbon – a reactive element

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Carbon – a reactive element

POE demonstrations – Student worksheets (Predict, Observe, Explain)

Activity 1: Blowing out candles

Your teacher will conduct this demonstration for you under safe laboratory conditions.



Alternatively, you can see this experiment in action here: http://www.trueactivist.com/15-awesome-chemistry-gifs/



Predict

What do you think will happen to the candle when the vinegar and baking soda mixture is poured over it?

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Observe Remember to use as many of your 5 senses (see, hear, touch, smell, taste) as is safe to use to observe what happens. What happened to the candle when the vinegar and baking soda mixture was poured over it?





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Activity 2: Dry ice bubbles

Your teacher will conduct this demonstration for you under safe laboratory conditions.



Alternatively, you can see this experiment in action here: http://www.youtube.com/watch?v=76CNkxizQuc



Predict

What do you think will happen when the block of dry ice is submerged into the warm soapy water?





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Observe

Remember to use as many of your 5 senses (see, hear, touch, smell, taste) as is safe to use to observe what happens.

What happened when the dry ice was submerged under the warm soapy water?







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Explain	; i !
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Activity 3: Sugar snake

Your teacher will conduct this demonstration for you under safe laboratory conditions.



Alternatively, you can see this experiment in action here: http://www.youtube.com/watch?v=AP6rTJi59NM



Predict

What do you think will happen when the acid is mixed with the sugar?









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Observe

Remember to use as many of your 5 senses (see, hear, touch, smell, taste) as is safe to use to observe what

What happened when the acid was mixed with the sugar?





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Explain
Can you explain what you just observed?
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Student observation activity: Where on earth is carbon?

Aim: To observe, compare and analyse a number of different forms of carbon.

SAFETY:

DO NOT taste or directly touch any of the carbon specimens.

Labels (and HAZCHEM warnings, where relevant) have been provided for each sample, giving additional safety instructions where appropriate.

Materials:

- > A small labelled* sample of each of the following substances: carbon dioxide, carbon monoxide, a piece of charcoal, a rod of carbon black, crude oil and chalk.
 - *This label will include a HAZCHEM sign (and accompanying explanation of its meaning) for any substances that are hazardous for humans.
- A labelled 3D molecular model of each of the following substances: graphite, diamond, Buckminster fullerene and carbon nanotube.

Method:

- 1. Ten different substances have been laid out around the room for you to observe**. Some of these are samples of the actual substance, while for others, their molecular structure is provided.
 - These substances include: carbon dioxide, carbon monoxide, a piece of charcoal, some carbon black powder, graphite, diamond, Buckminster fullerene, carbon nanotube, crude oil and chalk.
 - **Alternatively, you can view an image of each substance or molecular model on the following page.
- 2. Noting safety instructions above, carefully observe each of the specimens using as many of your senses as is safe and appropriate to do so.
 - If you are not sure how to approach the sample safely, please check with your teacher.
- 3. From your observations, complete as much of the table in the Results section as you can.
- 4. Conduct research (e.g. ask other students, look on websites, encyclopaedias, etc.) to find out the information for any sections of the results table that you are unable to complete yourself.

1. Carbon dioxide



2. Carbon monoxide



3. Charcoal









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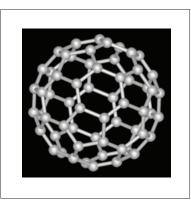
4. Carbon black



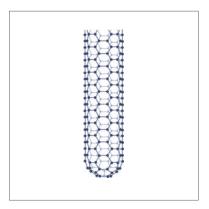
5. Diamond



6. Buckminster fullerene



7. Carbon nanotube



8. Graphite



9. Crude oil



10. Chalk



Sources:

- 1. http://i.ebayimg.com/00/s/MTAwMFgxMDAw/\$%28KGrHqV,!nME8V%2858tVSBPNJB5iljg
- ~~60_35.JPG 2. http://www.themakeupfoundation.com.au/media/pics/site/imagecache/5/6/562A60329 F2003A32506E2C5FA67EF64.jpg
 3. http://upload.wikimedia.org/wikipedia/commons/thumb/1/11/Charcoal-at-Campfire-
- DSC 3206w.jpg/760px-Charcoal-at-Campfire-DSC 3206w.jpg 4. http://upload.wikimedia.org/wikipedia/commons/c/c2/C%2C6.jpg
- $5.\,\underline{External\,appearance:}\,http://upload.wikimedia.org/wikipedia/commons/6/6f/Brillanten.jpg$

Structural image: http://upload.wikimedia.org/wikipedia/commons/9/96/Diamond animation.gif
6. http://upload.wikimedia.org/wikipedia/commons/8/82/Fullerene-C60.png

- 7. http://upload.wikimedia.org/wikipedia/commons/thumb/9/9b/Carbon_nanotube.svg/159px-Carbon_nanotube.svg.png
- 8. http://upload.wikimedia.org/wikipedia/commons/f/f3/GraphiteUSGOV.jpg 9. http://upload.wikimedia.org/wikipedia/commons/a/ae/Petroleum_cm05.jpg
- 10. http://upload.wikimedia.org/wikipedia/commons/1/14/Craie2.jpg









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Results:

Form of carbon

Answer the following 5 questions for each form of carbon in the table below.

- Q1: Observations (including state of matter at room temperature, colour and texture)
- Q2: What other elements (if any) are a part of this form?
- Q3: Chemical formula
- Q4: Where can this form of carbon be found on the Earth?
- Q5: What processes on the Earth (including man-made processes in the science lab) produce this form of carbon?

1. Carbon dioxide	Q1: Answer	Q2: Answer	Q3: Answer
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	Q4 : Answer	Q5: Answer	Other observations:
	4. Aliswei	G. Allswei	Other observations.
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2. Carbon monoxide	Q1: Answer	Q2: Answer	Q3: Answer
	Q4: Answer	Q5 : Answer	Other observations:
	 	1 1 1 1	





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3. Charcoal	Q1: Answer	Q2: Answer	Q3: Answer
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	Q4: Answer	Q5: Answer	Other observations:
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4. Carbon black	Q1: Answer	Q2: Answer	Q3: Answer
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	Q4: Answer	Q5: Answer	Other observations:
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5. Diamond	Q1: Answer	Q2: Answer Q5: Answer	Q3: Answer Other observations:
		QJ. AlliSWCI	Other observations.
6. Buckminster fullerene	Q1 : Answer	Q2: Answer	Q3: Answer
	Q4 : Answer	Q5 : Answer	Other observations:





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7. Nanotube	Q1: Answer	Q2: Answer	Q3: Answer
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	Q4: Answer	Q5: Answer	Other observations:
			!
8. Graphite	Q1: Answer	Q2: Answer	Q3: Answer
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	Q4: Answer	Q5: Answer	Other observations:
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9. Crude oil	Q1: Answer	Q2: Answer	Q3: Answer
	Q4: Answer	Q5: Answer	Other observations:
10. Chalk	Q1: Answer	Q2: Answer	Q3: Answer
		 	
	Q4: Answer	Q5: Answer	Other observations:
		, 	





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	so, identify them.
Q2	: There are many categories under which these substances can be grouped. For example, each substance
	exists naturally in one of three states of matter – solid, liquid and gas. Therefore, the 10 different sample could be grouped under these three categories.
	Devise one other way that you could place the 10 different samples into categories and draw this as a fin the space below.
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